



ISSN : 2350-0743

www.ijramr.com



International Journal of Recent Advances in Multidisciplinary Research

Vol. 02, Issue 12, pp.1060-1062, December, 2015

RESEARCH ARTICLE

BIODEGRADATION OF POLY (ETHYLENE TEREPHTHALATE) BY *BACILLUS SUBTILIS*

¹Asmae NAKKABI, ¹Nouredine ELMOUALIJ, ²Moulay SADIKI, ^{2,3}Saâd IBNSOUDA KORAICHI and ¹Mohammed FAHIM

¹Laboratory of Chemistry of Materials and Biotechnology Natural Products, Faculty of Sciences, University Moulay Ismail, BP 11201, 50000 Meknès, Morocco

²Laboratory of Microbial Biotechnology, Faculty of Science and Technology, 30000Fez Morocco

³Regional University Center of Interface, University Sidi Mohamed Ben Abdellah, 30000Fez Morocco

ARTICLE INFO

Article History:

Received 25th, September 2015

Received in revised form

26th, October 2015

Accepted 29th, November 2015

Published online 30th, December 2015

Keywords:

Biodegradation,
PET,
Bacillus Subtilis,
Polymer.

ABSTRACT

With the rapid development of poly (ethylene terephthalate) (PET) and its wide use in industry, there is a substantial fraction by volume added to the waste stream every year, which causes an environmental issue, since PET is highly resistant to atmospheric and biological agents. Currently, the handling methods of PET and other polymer wastes involve burying, burning, and recycling. But, as regards environmental protection, these methods have many shortcomings and cannot address poly (ethylene terephthalate) PET waste pollution from the source. Although the biodegradation ratio of PET was still weak, we demonstrated with SEM micrographs and IR analysis that *Bacillus subtilis* could act on the PET.

INTRODUCTION

Polymeric materials have gained a wide influence due to their excellent mechanical and thermal properties and high stability. They are very unique in chemical composition, physical forms, mechanical properties and applications. Because of this structural versatility, polymeric materials are widely used in aerospace applications, aviation and space industries as paints, adhesives, plastics, composites, rubbers, lubricants, etc (Vassiliou *et al.*, 2010). Environmental pollution by Polymeric materials, such as waste plastics, has been recognized as a large problem. In order to support continued sustainable development throughout the world, this problem must be addressed. In view of this, the biodegradation of plastics has been studied extensively for the past three decades. Some types of plastic have been shown to be biodegradable, and their degradation mechanisms have progressively become clearer (Masayuki, 2001). The biodegradation has the potential to avoid secondary pollution, while lowering the handling cost. This work has chosen as polymer poly(ethylene terephthalate). With the rapid development of poly(ethylene terephthalate) (PET) and its wide use in industry, there is a substantial fraction by volume added to the waste stream every year, which causes an environmental issue, since PET is highly resistant to atmospheric and biological agents.

Therefore, PET is a noxious material from a global environmental and ecological standpoint (Edge *et al.*, 1991; Allen *et al.*, 1994; Rudakova *et al.*, 1979).

MATERIALS AND METHODS

Polymer studied: polyester is a polymer (a chain of repeating units) where the individual units are held together by ester linkages. The specific polyester used in this work is polyethylene terephthalate (Fig 1).

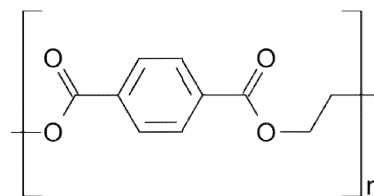


Fig. 1. The structure of its monomer (n: is the number of repetitions)

Organism: *Bacillus subtilis* was isolated, identified and obtained from the Laboratory of Microbial Biotechnology of Faculty of Science and Technology, Fez, by its ability to degrade a many polymers.

Degradation of Polyurethanes test, media and culture conditions: Luria-Bertani (LB) medium was prepared by adding 0.5g yeast extract, 10.NaCl, and 1.0g tryptone to 100 ml dH₂O. PET sample was also added to the medium.

*Corresponding author: ¹Asmae NAKKABI,

Laboratory of Chemistry of Materials and Biotechnology natural products,
University of Sciences Moulay Ismail, BP 11201, 50000Meknès, Morocco.

The test of biodegradation was incubated 30 days at 37°C IR analysis of polyurethane degradation

Scanning electron microscopy: SEM was used to characterize surface modifications of PET samples. Images of experimental surfaces showed decomposition and penetration of bacterium on the PET samples.

Infra red: IR spectra of the PET samples were analyzed using an IR spectrometer VERTEX 70, after 30 days of incubation.

RESULTS AND DISCUSSION

Surface morphology after biodegradation

The environmental scanning electron microscopy analysis before and after degradation of PET. The results of the effect of bacterial degradation on PET surface are presented in Fig 2. According to these findings, the surface of the sample control of PET is opaque and very smooth (Fig 2.A), compared to the samples which have subjected to the treatment by bacterial solutions. Indeed, the samples of PET polymer obtained after the immersion in the bacterial medium of *Bacillus subtilis* showed the whitish cracks formation on its surface (Fig 2.B).

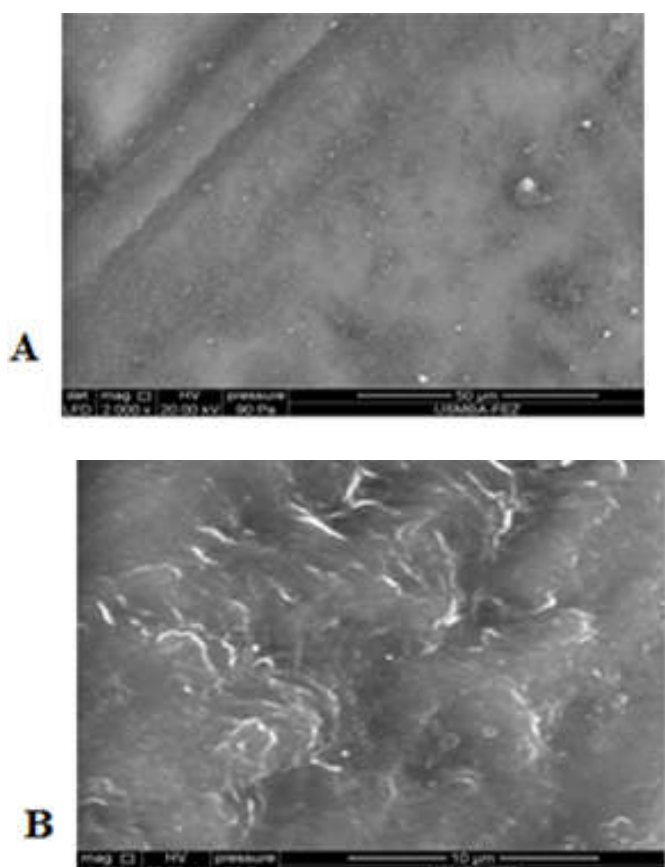


Fig. 2. (A) SEM images of PET before microbiological testing (B) SEM images of PET treated with *Bacillus subtilis*

From these results given by ESEM, it is found and noted that *Bacillus subtilis* strain has shown a significant interaction and effect on the PET surface producing low morphological heterogeneity and signs of erosion which show his ability to react and degrade the PET.

These results are consistent with a study of biodegradation of PET fibers and which also showed low degradation of this polymer (Jianfei *et al.*, 2004). Many researchers and institutes are engaged in the study of biodegradation of PET (Kohei, 2002; Huang *et al.*, 1980; Yoon *et al.*, 2001) but these studies are still in the primary stage due to the compact chemical structure of this polymer. PET spectrum is characterized by the presence of characteristic absorption bands of functions: (Fig 3A)

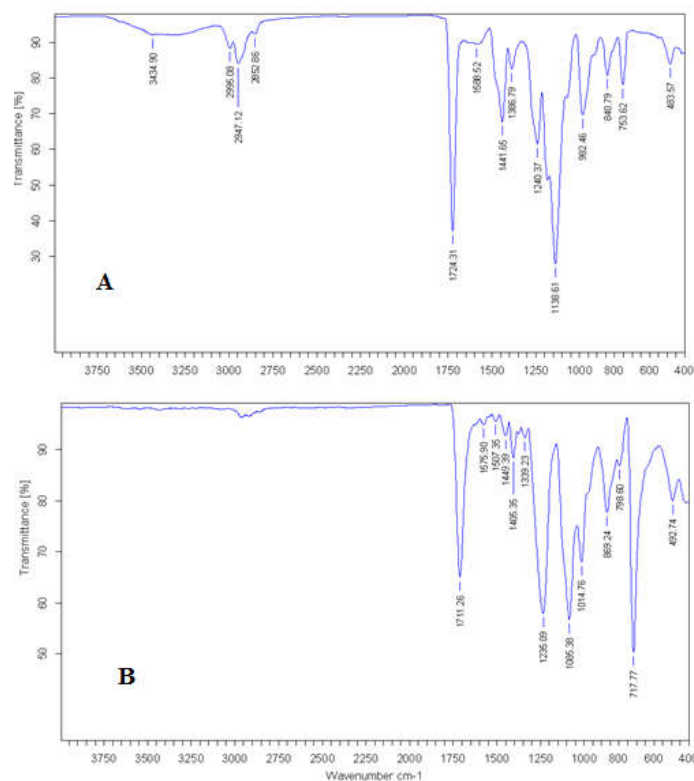


Fig. 3. (A): infrared spectrum of PET before test of biodegradation (B): infrared spectrum of PET after test of biodegradation

- Esters: 1050-1300 cm^{-1} (deformation $\delta_{\text{C-O}}$), 1720 to 1730 cm^{-1} (elongation $\nu_{\text{C=O}}$), 1240 cm^{-1} (deformation $\delta_{\text{O=C}}$);
- Aromatic at 1500-1600 cm^{-1} (deformation $\delta_{\text{C=C}}$) 3020 cm^{-1} (elongation $\nu_{\text{H-C}}$);
- l kyl at 1450 cm^{-1} (ν_{CH_2} elongation) and 715-750 cm^{-1} (deformation δ_{CH_2}).

The spectrum of the PET after biodegradation by *Bacillus* is characterized by the disappearance of absorption bands located at 2995, 2940, 2950 cm^{-1} (elongation $\nu_{\text{-CH}_2}$), decrease in intensity of the band located at 1711 cm^{-1} (ester carbonyl), the appearance of bands at 1339 and 1405 1449 cm^{-1} (deformation $\delta_{\text{-CH}_2}$), 1240, 1085 and 1014 cm^{-1} (elongation $\nu_{\text{-CO}_2}$) and 870 cm^{-1} (elongation $\nu_{\text{-CC}}$) (Fig 3B). In this study the degradation of PET by *Bacillus* has been chemically demonstrated by infrared spectroscopy, which shows A progressive reduction in the relative intensity of the peak carbonyl was observed and was accompanied by more subtle changes at another wave number (Fig 3). In the previous studies on activity of *Bacillus subtilis* toward different polymers (Nakkabi *et al.*, 2015). Such polyurethane were showed that *Bacillus subtilis* was able to degrade completely polyurethane.

The loss of this peak is consistent with hydrolysis of the ester bond in the urethane linkage (Nakkabi *et al.*, 2015). The growth of *Bacillus subtilis*, scattered on the surface of PET film, resulted in the emergence of a few small holes after one month (Fig 2A).

These results are the same given by other study who used the fungus *Penicillium funiculosum* (Bozena *et al.*, 2011) The current literature reports some fungi having the ability to degrade polyethylene terephthalate. This is the first study that demonstrates a capacity of the degradation of polyethylene terephthalate by *Bacillus subtilis*. Many researchers and institutes are engaged in the study of biodegradation of PET (Kohei, 2002; Huang *et al.*, 1980; Yoon *et al.*, 2001) but these studies are still in the primary stage due to the compact chemical structure of this polymer.

Conclusion

Widespread studies on the biodegradation of plastics have been carried out in order to overcome the environmental problems associated with synthetic plastic waste. Biodegradation of plastics by microorganisms and enzymes seems to be the most effective process to fight against plastic waste.

REFERENCES

- Allen, N. S., Edge, M., Mohammadian, M. and Jones, K. 1994. *Polym Degrad Stabil.* 43: 229.
- Edge, M., Hayes, M., Mohammadian, M., Allen, N.S., Jewitt, K., Brems, K. and Jones, K. 1991. *Polym Degrad Stabil.* 32: 131.
- Huang, S. J. and Byrne, C. A. J. 1980. *Appl. Polym. Sci.*, 25: 1951.
- Kohei, O. 2002. US 6.376, 213 B1.
- Nakkabi, A., Sadiki, M., Saad Ibnssouda and Fahim, M. 2015. Biological degradation of polyurethane by a newly isolated wood bacterium. *International Journal of Recent Advances in Multidisciplinary Research*, 02: 0222-0225.
- Nakkabi, A., Sadiki, M., Fahim, M., Ittobane, N. Koraichi, S.I., Barkai, H. and El Abed, S. 2015. Biodegradation of poly(ester urethane)s by *Bacillus subtilis*. *Int J Environ Res.*, 9: 157–162.
- NOWAK, B. et al. 2011. Biodegradation of poly(ethylene terephthalate) modified with polyester “Bionolle” by *Penicillium funiculosum*. *POLYMERY.* 56, nr 1.
- Rudakova, T. E., Zaikov, G.E., Voronkova, O.S., Daurova, T.T. and Degtyareva, S. M. 1979. *Sci. Polym. Symp.*, 66: 277.
- Shimao, M. 2001. Biodegradation of plastics. *Current Opinion in Biotechnology.* 12: 242–247.
- Vassiliou A.A., K. Chrissafis and Bikiaris, D.N. 2010. *Thermochimica Acta* 500. 1: 21–29.
- Yoon, M. Y., Kellis, J. and Poulou, A. J. 2001. Proceedings of the AATCC. International conference et exhibition AATCC. *Research Triangle Park* NC, CD-ROM.
- Zhang, J., Wang, X., Gong, J. and Gu, Z. 2004. A Study on the Biodegradability of Polyethylene Terephthalate Fiber and Diethylene Glycol Terephthalate. *Journal of Applied Polymer Science*, 93: 1089–1096.
